# Shri Rawatpura Sarkar University, Raipur



## **Examination Scheme & Syllabus**

## For

## **Bachelor of Pharmacy**

# Semester-II

(Effective from the session: 2019-20)



### Faculty of Pharmacy, Shri Rawatpura Sarkar University, Raipur Bachelor of Pharmacy

Semester-II Examination Scheme

### (Effective from the session: 2019-20)

#### SCHEME OF TEACHING AND EXAMINATION Table-I: Course of study for semester II

Sr. No.	Subject Code	Name of the Course with PCI code	Iı	nternal	assessme	nt					End s	semest	Total Marks	
			TA	Ses	sional exa	ams	Te	eachi	ing	Credit				
				СТ	Duration	Total	ho	ours j weel	per k					
							L	T P			Ma	arks	Duration	
1	BPH201T	Human Anatomy and Physiology II- Theory	10	15	1 Hr	25	3	1		4	75	25	3 Hrs	100
2	BPH202T	Pharmaceutical Organic Chemistry I-Theory		15	1 Hr	25	3	1		4	75	25	3 Hrs	100
3	BPH203T	Biochemistry-Theory	10	15	1 Hr	25	3	1		4	75	25	3 Hrs	100
4	BPH204T	Pathophysiology-Theory	10	15	1 Hr	25	3	1		4	75	25	3 Hrs	100
5	BSCPH205T	Computer Applications in Pharmacy *-Theory	10	15	1 Hr	25	2	1		3	50	25	2 Hrs	75
6	BSCPH206T	Environmental Studies-Theory	10	15	1 Hr	25	2	1		3	50	25	2 Hrs	75
7	BPH201P	Human Anatomy and Physiology II- Practical	05	10	4 Hrs	15			4	2	35	15	4 Hrs	50
8	BPH202P	Pharmaceutical Organic Chemistry I-Practical	05	10	4 Hrs	15			4	2	35	15	4 Hrs	50
9	BPH203P	Biochemistry- Practical	05	10	4 Hrs	15			4	2	35	15	4 Hrs	5 0
10	BSCPH205P	Computer Applications in Pharmacy- Practical *	05	05	2 Hrs	10			2	1	15	10	2 Hrs	25
			80	125	20 Hrs	205		Credits: 29			520		30 Hrs	725

\* The subject experts at college level shall conduct examinations.



#### **B. PHARMACY SEMESTER II SYLLABUS**

Course Title	H	Human Anatomy and Physiology II-Theory								
Course Code	Bl	PH2	201T	Total the	eory periods : 45 Hrs	Total Tutorial periods : 15				
Course	L	Т	Р	Credits	Total marks in the end semester : 75					
Credits	3	1		4	Minimum of class tests to be conducted : 02					
Prerequisites	Human Anatomy and Physiology I studied in previous class									
Course objectives	<ul> <li>Upon completion of this course the student should be able to:</li> <li>1. Explain the gross morphology, structure and functions of various organs of the human body.</li> <li>2. Describe the various homeostatic mechanisms and their imbalances.</li> <li>3. Identify the various tissues and organs of different systems of human body.</li> <li>4. Appreciate coordinated working pattern of different organs of each system</li> </ul>									
Course Contents	Unit I       • Body fluids and blood       10 hours         • Body fluids, composition and functions of blood, hemopoeisis, formation of haemoglobin, anemia, mechanisms of coagulation, blood grouping, Rh factors, transfusion, its significance and disorders of blood, Reticule endothelial system.         • Lymphatic system									

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Lymphatic organs and tissues, lymphatic vessels, lymph circulation and functions of lymphatic system Unit II 10 hours • Cardiovascular system Heart – anatomy of heart, blood circulation, blood vessels, structure and functions of artery, vein and capillaries, elements of conduction system of heart and heart beat, its regulation by autonomic nervous system, cardiac output, cardiac cycle. Regulation of blood pressure, pulse, electrocardiogram and disorders of heart. **Unit III** 06 hours • Digestive system Anatomy of GI Tract with special reference to anatomy and functions of stomach, (Acid production in the stomach, regulation of acid production through parasympathetic nervous system, pepsin role in protein digestion) small intestine and large intestine, anatomy and functions of salivary glands, pancreas and liver, movements of GIT, digestion and absorption of nutrients and disorders of GIT. • Respiratory system Anatomy of respiratory system with special reference to anatomy of lungs, mechanism of respiration, regulation of respiration Unit IV 10 hours • Respiratory system Lung Volumes and capacities transport of respiratory gases, artificial respiration, and resuscitation methods. • Urinary system Anatomy of urinary tract with special reference to anatomy of kidney and nephrons, functions of kidney and urinary tract, physiology of urine formation, micturition reflex and role of kidneys in acid base balance, role of RAS in kidney and disorders of kidney.

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	Unit V09 hours							
	Reproductive system							
	Anatomy of male and female reproductive system, Functions of male and female reproductive system, sex hormones, physiology of menstruation, fertilization, spermatogenesis, oogenesis, pregnancy and parturition							
	• Introduction to genetics							
	Chromosomes, genes and DNA, protein synthesis, genetic pattern of inheritance							
Course	1. Students would have studied about the gross morphology, structure and functions of nervous, respiratory, urinary and reproductive systems in the human body. Learn about visual activity, sense, auditory, differential WBC etc.							
outcomes	2. They would have studied in detailed about energy and metabolism.							
	3. Students would able to identify the various organs of different systems of human body.							
	1. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.							
Text books	2. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje ,Academic Publishers Kolkata							
Text Dooks	3. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brother's medical publishers, New Delhi.							
<b>Reference</b> books	1. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York.							
	<ol> <li>Physiological basis of Medical Practice-Best and Tailor. Williams &amp; Wilkins Co, Riverview, MI USA. 1.</li> </ol>							
	3. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co,							

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Riverview, MI USA

Course Title	Pl	Pharmaceutical Organic Chemistry I-Theory									
Course Code	B	PH	202	2T	Total theory periods : 45 Hrs	Total Tutorial periods : 15					
Course	L	Т	Р	Credits	Total marks in the end semeste	r : 75					
Credits	3	1		4	Minimum of class tests to be con	nducted : 02					
Prerequisite s	Ba	Basic common fundamental studied in pharmaceutical chemistry in diploma in pharmacy									
Course objectives		Upon completion of the course the student shall be able to 1. write the structure, name and the type of isomerism of the organic compound 2. write the reaction, name the reaction and orientation of reactions 3. account for reactivity/stability of compounds, 4. identify/confirm the identification of organic compound									
Course Contents	( ( ; ; ]	General methods of preparation and reactions of compounds superscripted with asterisk (*) to be explained To emphasize on definition, types, classification, principles/mechanisms, applications, examples and differences UNIT-I Hours 07									

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#### Bachelor of Pharmacy Semester-II 2023-24

nic compoun nerism's in org	on of Organic Compounds Common and IUPAC systems of nomenclatur ads (up to 10 Carbons open chain and carbocyclic compounds) Struc ganic compounds
UNIT-II	10 He
• Alkane	s*, Alkenes* and Conjugated dienes*
SP3 hybridiza SP2 hybridiza	ation in alkanes, Halogenation of alkanes, uses of paraffins. Stabilities of alke ation in alkenes
E1 and E2 carbocations, and $E_2$ react orientation, fr	reactions – kinetics, order of reactivity of alkyl halides, rearrangemen Saytzeffs orientation and evidences. $E_1$ verses $E_2$ reactions, Factors affectin tions. Ozonolysis, electrophilic addition reactions of alkenes, Markownike ree radical addition reactions of alkenes, Anti Markownikoff's orientation.
Stability of conjugated	onjugated dienes, Diel-Alder, electrophilic addition, free radical addition react l dienes, allylic rearrangement.
UNIT-III	10 Hot
• Alkyl h	alides*
SN1 and SN rearrangemen	2 reactions - kinetics, order of reactivity of alkyl halides, stereochemistry at of carbocations.
SN1 and SN rearrangemen SN1 versus S	<ul><li>2 reactions - kinetics, order of reactivity of alkyl halides, stereochemistry at of carbocations.</li><li>N2 reactions, Factors affecting SN1 and SN2 reactions</li></ul>
SN1 and SN rearrangemen SN1 versus S Structure an dichlorometh	<ul> <li><sup>12</sup> reactions - kinetics, order of reactivity of alkyl halides, stereochemistry at of carbocations.</li> <li><sup>10</sup> N2 reactions, Factors affecting SN1 and SN2 reactions</li> <li><sup>11</sup> uses of ethylchloride, Chloroform, trichloroethylene, tetrachloroethyl ane, tetrachloromethane and iodoform.</li> </ul>
SN1 and SN rearrangemen SN1 versus S Structure an dichlorometh • Alcohol alcohol	<ul> <li><sup>12</sup> reactions - kinetics, order of reactivity of alkyl halides, stereochemistry at of carbocations.</li> <li><sup>10</sup> N2 reactions, Factors affecting SN1 and SN2 reactions</li> <li><sup>11</sup> uses of ethylchloride, Chloroform, trichloroethylene, tetrachloroethyl ane, tetrachloromethane and iodoform.</li> <li><sup>11</sup> Is*- Qualitative tests, Structure and uses of Ethyl alcohol, chlorobutanol, Cetos, Benzyl alcohol, Glycerol, Propylene glycol</li> </ul>

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	Nucleophilic addition, Electromeric effect, aldol condensation, Crossed Aldol condensation, Cannizzaro reaction, Crossed Cannizzaro reaction, Benzoin condensation, Perkin condensation, qualitative tests, Structure and uses of Formaldehyde, Paraldehyde, Acetone, Chloral hydrate, Hexamine, Benzaldehyde, Vanilin, Cinnamaldehyde.
	UNIT-V 08 Hours
	Carboxylic acids*
	Acidity of carboxylic acids, effect of substituents on acidity, inductive effect and qualitative tests for carboxylic acids ,amide and ester
	Structure and Uses of Acetic acid, Lactic acid, Tartaric acid, Citric acid, Succinic acid. Oxalic acid, Salicylic acid, Benzoic acid, Benzyl benzoate, Dimethyl phthalate, Methyl salicylate and Acetyl salicylic acid
	• Aliphatic amines* - Basicity, effect of substituent on Basicity. Qualitative test, Structure and uses of Ethanolamine, Ethylenediamine, Amphetamine
Course outcomes	Understand the preparation, properties, general reaction and mechanism of basic reactions like substitution, elimination, addition etc. To perform common laboratory techniques including reflux, distillation, recrystallization, vacuum filtration, etc
	1. Organic Chemistry by P.L.Soni
Text books	2. Practical Organic Chemistry by Mann and Saunders.
	3. Vogel's text book of Practical Organic Chemistry
Reference books	<ol> <li>Organic Chemistry by Morrison and Boyd</li> <li>Organic Chemistry by I.L. Finar , Volume-I</li> </ol>
	3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.

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<b>Course Title</b>	Biochemistry-Th	eory								
Course Code	BPH203T	Total theory periods : 45 Hrs	Total Tutorial periods : 15							
Course	L T P Credits	Total marks in the end semeste	Total marks in the end semester : 75							
Credits	3 1 4	Minimum of class tests to be co	nducted : 02							
Prerequisites	Basic fundamental studied in previous chemistry & diploma course									
Course objectives	<ul> <li>Upon completion of course student shell able to</li> <li>1. Understand the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes.</li> <li>2. Understand the metabolism of nutrient molecules in physiological and pathological conditions.</li> <li>3. Understand the genetic organization of mammalian genome and functions of DNA in the synthesis of RNAs and proteins.</li> </ul>									
	<ul> <li>Biomolecules         <ul> <li>Biomolecules</li> <li>Introduction, classification, chemical nature and biological role of carbohydrate, lipids, nucleic acids, amino acids and proteins.</li> <li>Bioenergetics</li> <li>Concept of free energy, endergonic and exergonic reaction, Relationship between free energy, enthalpy and entropy; Redox potential. Energy rich compounds; classification; biological significances of ATP and cyclic AMP</li> <li>UNIT II</li> </ul> </li> </ul>									
	<ul> <li>Carbohydrate metabolism</li> <li>Glycolysis – Pathway, energetics and significance Citric acid cycle- Pathway, energetics and significance HMP shunt and its significance; Glucose-6-Phose dehydrogenase (G6PD) deficiency</li> <li>Glycogen metabolism Pathways and glycogen storage diseases (GSD)</li> <li>Gluconeogenesis-Pathway and its significance Hormonal regulation of blood glucose level and Diabetes mellitu</li> </ul>									
	• Biological	oxidation								
	Electron transport chain (ETC) and its mechanism. Oxidative phosphorylation & its mechanism and substrate level phosphorylation Inhibitor ETC and oxidative phosphorylation/Uncouplers									

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	UNIT III 10 Hours
	• Lipid metabolism
	$\beta$ -Oxidation of saturated fatty acid (Palmitic acid)
Course	<ul> <li>Formation and utilization of ketone bodies; ketoacidosis De novo synthesis of fatty acids (Palmitic acid) Biological significance of cholesterol and conversion of cholesterol into bile acids, steroid hormone and vitamin D</li> <li>Disorders of lipid metabolism: Hypercholesterolemia, atherosclerosis, fatty liver and obesity.</li> <li>Amino acid metabolism</li> <li>General reactions of amino acid metabolism: Transamination, deamination &amp; decarboxylation, urea cycle and its disorders.</li> <li>Catabolism of phenylalanine and tyrosine and their metabolic disorders (Phenyketonuria, Albinism, alkeptonuria, tyrosinemia)</li> <li>Synthesis and significance of biological substances; 5-HT, melatonin, dopamine, noradrenaline, adrenaline Catabolism of heme; hyperbilirubinemia and jaundice</li> </ul>
	• Nucleic acid metabolism and genetic information transfer Biosynthesis of purine and pyrimidine nucleotides Catabolism of purine nucleotides and Hyperuricemia and Gout disease Organization of mammalian genome
Contents	Structure of DNA and RNA and their functions DNA replication (semi conservative
	model) Transcription or RNA synthesis
	Genetic code, Translation or Protein synthesis and inhibitors
	UNIT IV 08 Hours
	Biomolecules
	<ul> <li>Introduction, classification, chemical nature and biological role of carbohydrate, lipids, nucleic acids, amino acids and proteins.</li> <li>Bioenergetics</li> <li>Concept of free energy, endergonic and evergonic reaction. Relationship between free</li> </ul>
	<ul> <li>Concept of free energy, endergonic and exergonic freaction, Kerationship between free energy, enthalpy and entropy; Redox potential.</li> <li>Energy rich compounds; classification; biological significances of ATP and cyclic AMP UNIT V</li> <li>O7 Hours</li> <li>Enzymes</li> <li>Introduction, properties, nomenclature and IUB classification of enzymes Enzyme kinetics</li> </ul>
	(Michaelis plot, Line Weaver Burke plot)
	Enzyme inhibitors with examples Regulation of enzymes: enzyme induction and repression, allosteric enzymes regulation

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	Therapeutic and diagnostic applications of enzymes and isoenzymes Coenzymes –Structure and biochemical functions.
	1. Understand the knowledge of biochemical reactions and pathways and efficiency diseases, structures functions of DNA and RNA.
Course	2. To acquire knowledge in qualitative and quantitative estimation of the biological macromolecules.
outcomes	3. To know the interpretation of data emanating from a Clinical Test Lab.
	4. To know how physiological conditions influence the structures and re -activities of biomolecules
	5. To understand the basic principles of protein and polysaccharide structure.
Text books	<ol> <li>Textbook of Biochemistry by Rama Rao.</li> <li>Textbook of Biochemistry by Deb.</li> <li>Outlines of Biochemistry by Conn and Stumpf</li> <li>Practical Biochemistry by R.C. Gupta and S. Bhargavan.</li> </ol>
Reference books	<ol> <li>Principles of Biochemistry by Lehninger.</li> <li>Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell.</li> <li>Biochemistry by Stryer.</li> <li>Biochemistry by D. Satyanarayan and U.Chakrapani.</li> </ol>

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<b>Course Title</b>	Path	opł	nysiology-7	Гheory							
<b>Course Code</b>	BPH	204	4T	<b>Total theory periods : 45 Hrs</b>	Total Tutorial periods : 15						
Course	LT	P	Credits	Total marks in the end semester	: 75						
Credits	3 1		4	Minimum of class tests to be cond	ducted : 02						
Prerequisites	Anat	Anatomy and physiology detailed studied in previous HAP									
	U	pon	completion of	of the subject student shall be able to $-$							
Course	1.	1. Describe the etiology and pathogenesis of the selected disease states; 2. Name the signs and symptoms of the diseases; and									
objectives	2.	2. Name the signs and symptoms of the diseases; and 3 Mention the complications of the diseases									
	3.	3.Mention the complications of the diseases.									
		Un	nit I		10 Hours						
		• E	Basic princ	iples of Cell injury and Adaptation	n:						
		Int	roduction,	definitions, Homeostasis, Componer	nts and Types of Feedback systems,						
		Ca	uses of cell	lular injury,Pathogenesis (Cell memb	orane damage, Mitochondrial damage,						
		Ril	bosome dai	nage, Nuclear damage),Morphology	of cell injury – Adaptive changes						
		(Atrophy, Hypertrophy, hyperplasia, Metaplasia, Dysplasia), Cell swelling, Intra cellular									
		accumulation, Calcification, Enzyme leakage and Cell Death Acidosis									
		&Alkalosis,Electrolyte imbalance									
		• Basic mechanism involved in the process of inflammation and repair:									
		Introduction, Clinical signs of inflammation, Different types of Inflammation, Mechanism									
		of	Inflammati	on – Alteration in vascular permeable	ility and blood flow, migration of						
		W]	BC's,Medi	ators of inflammation, Basic principle	es of wound healing in the						
Course		ski	n,Pathophy	vsiology of Atherosclerosis							
Contents	Unit II 10 Hour										
	• Cardiovascular System:										
	Hypertension, congestive heart failure, ischemic heart disease (angina,myocardial										
		infarction, atherosclerosis and arteriosclerosis)									
		• F	Respiratory	system:Asthma, Chronic obstructive	e airways diseases.						
		• F	Renal system	n:Acute and chronic renal failure							
		Unit III 10 Hours									
		• †	laematolo	gical Diseases:	and falia anid) sights call anomia						
		Ir (1	on deficier	icy, megaloblastic anemia (vit B12 a	and folic acid), sickle cell anemia,						
		tha	liasemia, no	ereditary							
		a	equired ane	emia, hemophilia							
		• E	Endocrine	system: Diabetes, thyroid diseases, o	disorders of sex hormones.						
		• N	Nervous sy	stem: Epilepsy, Parkinson's disease,	, and stroke, psychiatric disorders:						

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	depression, schizophrenia
	and Alzheimer's disease.
	Gastrointestinal system: Peptic Ulcer
	Unit IV 08 Hours
	• Inflammatory bowel diseases, jaundice, hepatitis (A, B, C, D, E, F) alcoholic liver disease.
	• Disease of bones and joints: Rheumatoid arthritis, osteoporosis and gout
	• Principles of cancer: classification, etiology and pathogenesis of cancer
	Unit V 07 Hours
	• Infectious diseases: Meningitis, Typhoid, Leprosy, Tuberculosis Urinary tract infections
	Sexually transmitted diseases: AIDS, Syphilis, Gonorrhea
Course	<ol> <li>Describe the etiology and pathogenesis of the selected disease states</li> <li>Knowledge of signs and symptoms of the diseases 3. Identify the complications of the diseases.</li> </ol>
outcomes	4. Know most commonly encountered pathophysiological state(s) and/or disease mechanism(s), as well as any clinical testing requirements
Text books	<ol> <li>Vinay Kumar, Abul K. Abas, Jon C. Aster; Robbins &amp;Cotran Pathologic Basis of Disease; South Asia edition; India; Elsevier; 2014.</li> <li>William and Wilkins, Baltimore; 1991 [1990 printing].</li> </ol>
	<ol> <li>Harsh Mohan; Text book of Pathology; 6<sup>th</sup> edition; India; Jaypee Publications; 2010.</li> <li>Laurence B, Bruce C, Bjorn K.; Goodman Gilman's The Pharmacological Basis of</li> </ol>
Reference	Therapeutics; 12 <sup>th</sup> edition; New York; McGraw-Hill; 2011.
books	3. Best, Charles Herbert 1899-1978; Taylor, Norman Burke 1885-1972; West, John B (John Burnard); Best and Taylor's Physiological basis of medical practice; 12th ed; united states;

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Course Title	Computer Applications in Pharmacy-Theory										
Course Code	BSCPH2	05T Total t	theory periods : 30 Hrs	Total Tutorial periods : 10							
Course	LTP	Credits	Total marks in the end	d semester : 50							
Credits	2 1	3	Minimum of class tests	s to be conducted : 02							
Prerequisites	Basic cor	nmon fundan	nental studied in higher	secondary education							
	Upon completion of the course the student shall be able to										
Course	<ol> <li>know the various types of application of computers in pharmacy</li> <li>know the various types of databases</li> </ol>										
objectives											
	3. kno	ow the various	applications of databases	s in pharmacy							
	UNIT -	-I		06 hours							
	Numbe	er system: Bir	hary number system, Deci	mal number system, Octal number system,							
	Hexade	ecimal number	systems, conversion deci	imal to binary, binary to decimal, octal to binary							
	etc, bin	ary addition, t	Dinary subtraction – One's	s complement, I wo's complement method, binar	У						
	Concor	multiplication, binary division									
	fassibility analysis data flow diagrams process specifications input/output design process life										
	cycle, planning and managing the project										
	UNIT –II										
				06 hours							
	Web technologies : Introduction to HTML, XML, CSS and Programming languages, introduction										
	to web servers and Server Products Introduction to databases, MYSQL, MS ACCESS, Pharmacy										
Course	Drug da	atabase									
Contents	UNIT -	06 hours									
Contents	Applic	ation of comp	<b>Duters in Pharmacy</b> – Dr	ug information storage and retrieval,							
	Finarmacokinetics, Mathematical model in Drug design, Hospital and Clinical Pharmacy,										
	automated dispensing of drugs mobile technology and adherence monitoring										
	Diagnostic System I ab-diagnostic System Patient Monitoring System Pharma Informati										
	System	1									
	UNIT -	-IV		06 hours							
	Bioinfo	ormatics: Intro	oduction, Objective of Bi	oinformatics, Bioinformatics Databases, Concept	C						
	of Bioi	nformatics, In	pact of Bioinformatics in	Vaccine Discovery							
	UNIT-	V		06 hours							
	Compu	uters as data a	analysis in Preclinical de	evelopment: Chromatographic dada							
	analysi	s(CDS), Labor	ratory Information manag	ement System (LIMS) and Text Information							
	Manage	ement System	(TIMS)								
Course	1. Des	ign and develo	op solutions to analyze ph	armaceutical problems using computers.							
outcomes	2. Integ	grate and apply	y efficiently the contempo	prary IT tools to all Pharmaceutical related activit	ties						

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	3. Solve and work with a professional context pertaining to ethics, social, cultural and regulations
	with regard to
	Pharmacy.
	1. Bioinformatics (Concept, Skills and Applications) - S.C.Rastogi-CBS Publishers
	and Distributors, 4596/1- A, 11 Darya Gani, New Delhi – 110 002(INDIA)
Text books	2. Microsoft office Access - 2003, Application Development Using VBA, SQL Server,
	DAP and Infopath -Cary N.Prague - Wiley Dreamtech India (P) Ltd., 4435/7,
	Ansari Road, Daryagani, New Delhi - 110002
	1. Computer Application in Pharmacy – William E.Fassett –Lea and Febiger,
Reference	600 South Washington Square, USA, (215) 922-1330.
books	2. Computer Application in Pharmaceutical Research and Development –Sean
	Ekins – Wiley-Interscience, A John Willey and Sons, INC., Publication, USA

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<b>Course Title</b>	En	Environmental Studies-Theory						
Course Code	BSCPH206T				<b>Total theory periods : 30 Hrs</b>	<b>Total Tutorial periods : 10</b>		
Course Credits	L	Т	P	Details	Total marks in the end semester :	: 50		
	2	1		3	Minimum of class tests to be cond	lucted : 02		
Prerequisites	Ba	Basic common knowledge studied in higher secondary education						
Course objectives	<ul> <li>Upon completion of the course the student shall be able to <ol> <li>Create the awareness about environmental problems among learners</li> <li>Impart basic knowledge about the environment and its allied problems.</li> <li>Develop an attitude of concern for the environment.</li> <li>Motivate learner to participate in environment protection and environment improvement.</li> </ol> </li> <li>Acquire skills to help the concerned individuals in identifying and solving environmental problems.</li> <li>Strive to attain harmony with nature.</li> </ul>							
	Unit-I The Multidisciplinary nature of environmental studies Natural Resources Renewable and non-renewable resources: Unit-II Natural resources and associated problems a) Forest resources; b) Water resources; c) Mineral resources; d) Food resources; e) Energy resources: f) L and resources: Pole of an individual in conservation of natural resources							
Course Contents	Unit-III Ecosystems Concept of an ecosystem. Structure and function of an ecosystem. Unit-IV Introduction, types, characteristic features, structure and function of the ecosystems: Forest ecosystem; Grassland ecosystem; Desert ecosystem; Aquatic ecosystems (ponds, streams, lakes, rivers, oceans, estuaries) Unit-V							
Course		nvir ר	onn	nental Pollu	tion: Air pollution; Water pollution; So	bil pollution		
Course		. 1	1 M1S	program	snall create awareness about envir	ronmental problems; develop an		
outcomes		ä	uuu	ide towards				

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	2. Students gain knowledge about Enivronmental conditions, hazards, contaminants and
	their analysis.
	Recommended Books (Latest edition):
	1. Agarwal, K.C. 2001 Environmental Biology, Nidi Publ. Ltd. Bikaner.
Text books	2. Bharucha Erach, The Biodiversity of India, Mapin Pu blishing Pvt. Ltd., Ahmedabad –
	380 013, India,
	3. Cunningham, W.P. Cooper, T.H. Gorhani, E & Hepworth, M.T. 2001,
References books	1. Y.K. Sing, Environmental Science, New Age International Pvt, Publishers, Bangalore
	2. Brunner R.C., 1989, Hazardous Waste Incineration, McGraw Hill Inc. 480p
	3. Clark R.S., Marine Pollution, Clanderson Press Oxford.

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<b>Course Title</b>	Human Anatomy and Physiology II -Practical							
<b>Course Code</b>	BPH201P	Total Practical periods : 04 Hrs / week						
Course	L T P Credits	Total marks in the end semester : 35						
Credits	4 2							
Prerequisites	Student must have Basic knowledge of different system of body							
Course objectives	<ol> <li>Explain the gross morphology, structure and functions of various organs of the human body.</li> <li>Describe the various homeostatic mechanisms and their imbalances.</li> <li>Identify the various tissues and organs of different systems of human body.</li> <li>Appreciate coordinated working pattern of different organs of each system</li> </ol>							
Course Contents	<ul> <li>Practical physiology is complimentary to the theoretical discussions in physiology. Practicals allow the verification of physiological processes discussed in theory classes through experiments on living tissue, intact animals or normal human beings. This is helpful for developing an insight on the subject.</li> <li>1. Introduction to haemocytometer.</li> <li>2. Enumeration of white blood cell (WBC) count</li> <li>3. Enumeration of total red blood corpuscles (RBC) count</li> <li>4. Determination of loteding time</li> <li>5. Determination of clotting time</li> <li>6. Estimation of haemoglobin content</li> <li>7. Determination of blood group.</li> <li>8. Determination of heart rate and pulse rate.</li> <li>10. Recording of blood pressure.</li> <li>11. Determination of tidal volume and vital capacity.</li> <li>12. Study of digestive, respiratory, cardiovascular systems, urinary and reproductive systems with the help of models, charts and specimens.</li> <li>13. Recording of basal mass index</li> <li>14. Study of family planning devices and pregnancy diagnosis test.</li> <li>15. Demonstration of total blood count by cell analyser</li> <li>16. Permanent slides of vital organs and gonads.</li> <li>1. Students would have studied about the gross morphology, structure and functions of nervous, respiratory.</li> </ul>							
Course outcomes	<ol> <li>Students we nervous, respurinary and auditory, dif</li> <li>They would h</li> </ol>	build have studied about the gross morphology, structure and functions of piratory, reproductive systems in the human body. Learn about visual activity, sense, ferential WBC etc.						

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	3. Students would able to identify the various organs of different systems of human body.
Text books	<ol> <li>Physiological basis of Medical Practice-Best and Tailor. Williams &amp; Wilkins Co, Riverview, MI USA.</li> <li>Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brother's medical publishers, New Delhi.</li> </ol>
Reference books	1. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York.

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Course Title	Pharmaceutical Organic Chemistry I -Practical					
Course Code	BPH20	2P	Total Practical periods : 04 Hrs / week			
Course	L T P	Credits	Total marks in the end semester : 35			
Credits	4	2				
Prerequisites	Basic d	letails studi	ed in previous II sem. Pharmaceutical Organic Chemistry			
Course objectives	Upon c 1. 2. 3. 4.	completion of write the s write the re account fo identify/cc	the course the student shall be able to tructure, name and the type of isomerism of the organic compound eaction, name the reaction and orientation of reactions r reactivity/stability of compounds, nfirm the identification of organic compound			
Course Contents	<ol> <li>Systematic qualitative analysis of unknown organic compounds like         <ol> <li>Preliminary test: Color, odour, aliphatic/aromatic compounds, saturation and unsaturation, etc.</li> <li>Detection of elements like Nitrogen, Sulphur and Halogen by Lassaigne's test</li> <li>Solubility test</li> <li>Functional group test like Phenols, Amides/ Urea, Carbohydrates, Amines, Carboxylic acids, Aldehydes and Ketones, Alcohols, Esters, Aromatic and Halogenated Hydrocarbons, Nitro compounds and Anilides.</li> <li>Melting point/Boiling point of organic compounds</li> <li>Identification of the unknown compound from the literature using melting point/ boiling point</li> </ol> </li> </ol>					
Course Contents	7. Pr 8. M 9. Pro 10. C	reparation o inimum 5 u eparation of constructior	f the derivatives and confirmation of the unknown compound by melting point/ boi inknown organic compounds to be analysed systematically. If suitable solid derivatives from organic compounds of molecular models.			
Course outcomes	Unde substi reflux vacuu	rstand the j tution, elin , distillation m filtration	preparation, properties, general reaction and mechanism of basic reactions like nination, addition etc. To perform common laboratory techniques including n, recrystallization, , etc			

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Text books	<ol> <li>Organic Chemistry by Morrison and Boyd</li> <li>Textbook of Organic Chemistry by B.S. Bahl &amp; Arun Bahl.</li> <li>Organic Chemistry by P.L.Soni</li> <li>Practical Organic Chemistry by Mann and Saunders.</li> </ol>
Reference books	<ol> <li>Organic Chemistry by I.L. Finar , Volume-I</li> <li>Vogel's text book of Practical Organic Chemistry</li> </ol>

Course Title	Bio	Biochemistry -Practical (BP209P)				
Course Code	BPH203P			Total Practical periods : 04 Hrs / week		
Course	L	ГР	Details	Total marks in the end semester : 35		
Credits		4	2			
Prerequisites	Bas	ic fı	indamenta	l studied in previous chemistry & diploma course		
	U	Jpor	o completio	n of course student shell able to		
Course objectives	1 2 3	<ol> <li>Understand the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes.</li> <li>Understand the metabolism of nutrient molecules in physiological and pathological conditions.</li> <li>Understand the genetic organization of mammalian genome and functions of DNA in the synthesis of RNAs and proteins.</li> </ol>				
Course Contents	1 2 3	<ol> <li>Qualitative analysis of carbohydrates (Glucose, Fructose, Lactose, Maltose, Sucrose and starch)</li> <li>Identification tests for Proteins (albumin and Casein)</li> <li>Quantitative analysis of reducing sugars (DNSA method) and Proteins (Biuret method)</li> </ol>				

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	4. Qualitative analysis of urine for abnormal constituents
	5. Determination of blood keratinises.
	6. Determination of blood sugar
	7. Determination of serum total cholesterol
	8. Preparation of buffer solution and measurement of pH
	9. Study of enzymatic hydrolysis of starch
	10.Determination of Salivary amylase activity
	11.Study the effect of Temperature on Salivary amylase activity.
	12.Study the effect of substrate concentration on salivary amylase activity.
	1. Understand the knowledge of biochemical reactions and pathways and efficiency diseases,
	structures functions of DNA and RNA.
	2. To acquire knowledge in qualitative and quantitative estimation of the biological
Course	macromolecules.
outcomes	3. To know the interpretation of data emanating from a Clinical Test Lab.
	4. To know how physiological conditions influence the structures and re -activities of
	biomolecules.
	5. To understand the basic principles of protein and polysaccharide structure.
	1. Biochemistry by Stryer.
Toyt books	2. Biochemistry by D. Satyanarayan and U.Chakrapani
I CAL DUUKS	3. Textbook of Biochemistry by Rama Rao.
	4. Textbook of Biochemistry by Deb.
Reference	1. Principles of Biochemistry by Lehninger.
books	2. Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell.

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<b>Course Title</b>	Computer Applications in Pharmacy-Practical (BP210P)					
Course Code	BSC	CPI	H 205P	Total Practical periods : 02 Hrs / week		
Course	LT	P	Credits	Total marks in the end semester : 15		
Credits		2	1			
Prerequisites	Basic	c co	mmon kno	wledge studied in higher secondary education		
Course objectives	Upo 1.ki 2.ki	Upon completion of the course the student shall be able to 1.know the various types of application of computers in pharmacy 2.know the various types of databases				
Course Contents	5. Ki part 2. ( 3. I 4. ( 5. ( 6. I 7. ( 8. ( 9. I 10. ( 11. ] 12. E	Des icu Crea Retr Crea Crea Gen Crea Cre	ign a ques ign a ques lar disease. ate a HTMI rieve the inf ating mailin ating mailin ating mailin ating mailin ating mailin ating mailin ating mailin ating mailin ating mailin ating and w porting Tables.	stionnaire using a word processing package to gather information about a L web page to show personal information. formation of a drug and its adverse effects using online tools and labels Using Label Wizard, generating label in MS WORD ase in MS Access to store the patient information with the required fields Using in MS Access to view, add, delete and modify the patient record in the database fort and printing the report from patient database to table using – MS Access on storage and retrieval using MS Access porking with queries in MS Access les, Queries, Forms and Reports to web pages Queries, Forms and Reports to XML pages		
Course outcomes	1. 2. 3. wi	De Inte Sol th r P	sign and de egrate and a ve and wor egard to harmacy.	velop solutions to analyze pharmaceutical problems using computers. pply efficiently the contemporary IT tools to all Pharmaceutical related activities k with a professional context pertaining to ethics, social, cultural and regulations		
Text books	1. 2.	Bic Dis Mic and Dar	oinformatics stributors, 4 crosoft offic l Infopath - ryagani, Ne	s (Concept, Skills and Applications) – S.C.Rastogi-CBS Publishers and 596/1- A, 11 Darya Gani, New Delhi – 110 002(INDIA) ce Access - 2003, Application Development Using VBA, SQL Server, DAP -Cary N.Prague – Wiley Dreamtech India (P) Ltd., 4435/7, Ansari Road, w Delhi 110002		

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Reference	1. Computer Application in Pharmacy – William E.Fassett –Lea and Febiger, 600 South Washington Square, USA, (215) 922-1330.
books	<ol> <li>Computer Application in Pharmaceutical Research and Development –Sean Ekins – Wiley-Inter science, A John Willey and Sons, INC., Publication, USA</li> </ol>

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